

Unit Plan: MATTER AND ITS INTERACTIONS

Grade: 7th

Subject: Physical Science

Term: 1

Thematic Concept: INTERACTIONS	Universal EQ: “WHY IS IT IMPORTANT TO UNDERSTAND INTERACTIONS?”	
Supporting Concept: HIERARCHY	Content-Based EQ: “HOW DO HIERARCHIES INFLUENCE INTERACTIONS?”	
Generalizations: Matter is not created nor destroyed.		
Standard(s): HS-PS1-1. Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level of atoms. MS-PS1-1. Develop models to describe the atomic composition of simple molecules and extended structures. MS-PS1-3. Gather and make sense of information to describe that synthetic materials come from natural resources and impact society. MS-PS1-4. Develop a model that predicts and describes changes in particle motion, temperature, and state of a pure substance when thermal energy is added or removed.		
Unit Outcomes: The emphasis of this unit is on the use of models and evidence to answer how atomic and molecular interactions explain properties of matter.		
Unit Description: This course focuses on atomic and molecular structure; properties of matter; physical and chemical properties; and state changes.		
Conceptual Knowledge Students will understand : <ul style="list-style-type: none"> • Substances’ structure follows its function or function follows its structure. • Matter makes up everything, is conserved, and goes through physical and chemical changes. • Science and engineering practices. 	Procedural Knowledge Students will be able to do : <ul style="list-style-type: none"> • How to implement safety procedures. • Separate homologous and heterogeneous mixtures. • Carry out various investigations to identify physical/chemical properties, and Boyles/ Charles’ Laws. • Explain how kinetic energy changes the state of matter. 	Factual Knowledge Students will know : <ul style="list-style-type: none"> • Laboratory safety procedures • Scientific method of inquiry • The structure, properties, and interactions of matter • Vocabulary related to matter • How to collect, record, and chart data • Pure substances have different physical and chemical properties
Conceptual Formative Assessments: <ul style="list-style-type: none"> • Properties of Matter Essential Question Reflection 	Procedural Formative Assessments: <ul style="list-style-type: none"> • Separation of Mixture Challenge <ul style="list-style-type: none"> ○ Use laboratory techniques to 	Factual Formative Assessments: <ul style="list-style-type: none"> • Quick Review

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<ul style="list-style-type: none"> ○ How do hierarchies influence interactions between different types of matter? There should be at least 3 examples (2 points each) from class used as evidence to support your claim (4 points). ○ Why is it important to understand the interaction between structure and function? 	<ul style="list-style-type: none"> ○ separate an unknown mixture. ○ Document the investigation, analyze, and interpret data. ● Common Labs (Need to decide on which ones we want to use!) 	
<p>Conceptual Summative Assessments:</p> <ul style="list-style-type: none"> ● Justify this statement: Matter makes up everything, is conserved, and goes through physical and chemical changes. ● How does substances' structure follows its function? OR How does function follows its structure? Use three examples from class using evidence to support your claim. 	<p>Procedural Summative Assessments:</p> <ul style="list-style-type: none"> ● Properties of Matter Unit Test 	<p>Factual Summative Assessments:</p> <ul style="list-style-type: none"> ● Properties of Matter Unit Test <ul style="list-style-type: none"> ○ (This is something we will have to make and make sure everyone is using it!)
<p>Affective Self-Regulatory Strategies:</p> <ul style="list-style-type: none"> ● Building confidence through collaboration 	<p>Behavioral Self-Regulatory Strategies:</p> <ul style="list-style-type: none"> ● Following the science safety procedures ● Scientific methods ● Time management ● Collaboration 	<p>Cognitive Self-Regulatory Strategies:</p> <ul style="list-style-type: none"> ● Reflection on daily collaboration ● Science and engineering practices
<p>Resources/Materials</p> <ul style="list-style-type: none"> ● Different materials for mixture lab, cups, scales ● Phase change lab materials ● Solutions, lasers, flashlights ● PASCO probes ● Supplies to build atoms and molecules 		

*THIS DOCUMENT ONLY CONTAINS COMMON FORMATIVE AND SUMMATIVE ASSESSMENTS.